

Al-B-Ho (Aluminum-Boron-Holmium)

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[1994Mik] determined an isothermal section at 600 °C for this ternary system, which depicts three ternary compounds.

Binary Systems

There are two intermediate phases in the Al-B system: AlB_2 ($C32$, AlB_2 -type hexagonal) and AlB_{12} (AlB_{12} -type tetragonal). The Al-Ho system [1988Gsc] depicts five intermediate phases: HoAl_3 ($hR20$, rhombohedral), HoAl_2 ($C15$, MgCu_2 -type cubic), HoAl (ErAl-type orthorhombic), Ho_3Al_2 ($Zr_3\text{Al}_2$ -type tetragonal), and Ho_2Al ($C23$, Co_2Si -type orthorhombic). The B-Ho phase diagram [Massalski2] shows the following compounds: HoB_2 ($C32$, AlB_2 -type

hexagonal), HoB_4 ($D1_c$, ThB_4 -type tetragonal), HoB_6 ($D2_1$, CaB_6 -type cubic), HoB_{12} ($D2_f$, UB_{12} -type cubic) and HoB_{66} (cubic).

Ternary Compounds

Three ternary compounds are known in this system. HoAl_3B_x ($x = 0.4-0.5$) (BaPb_3 -type rhombohedral, space group $R\bar{3}m$, denoted τ_1 here) [1980Mik], HoAlB_4 (YCrB_4 -type orthorhombic, space group $Pbam$, $a = 0.59283$ nm, $b = 1.1553$ nm, and $c = 0.35306$ nm, denoted τ_2 here) [1994Mik]; and HoAlB_{14} (MgAlB_{14} -type orthorhombic, space group $Imma$, denoted τ_3 here) [1989Kor].

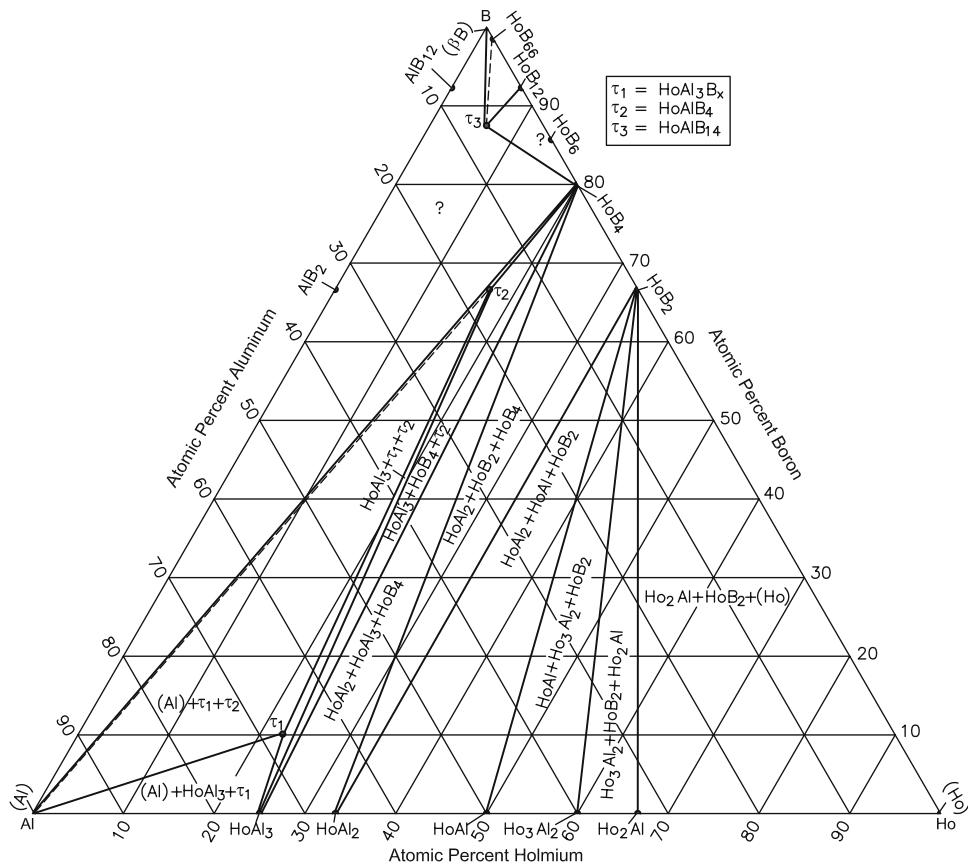


Fig. 1 Al-B-Ho isothermal section at 600 °C [1994Mik]. Thin two-phase regions are omitted

Ternary Isothermal Section

With starting materials of purity ≥ 99.5 mass% purity, [1994Mik] arc-melted or sintered 70 ternary alloys. The alloys were annealed at 600 °C for not less than 500 h and quenched in water. The phase equilibria were studied by x-ray powder diffraction. Single crystals grown from aluminum-rich melts were examined by the Laue method. The isothermal section at 600 °C constructed by [1994Mik] is shown in Fig. 1. All three ternary compounds are present at this temperature. The phase equilibria involving AlB₂, AlB₁₂ and HoAlB₁₄ (τ_3) were not determined by [1994Mik]. The binary phase HoB₆ does not appear in the isothermal section given by [1994Mik].

References

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